## LIQUID CRYSTALS

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# Identification of Nematic Lyomesophases with Discotic and Cylindrical Micelles in Lyotropic Amphiphilic Systems

E. O. Kiĭrend<sup>a</sup>, S. P. Chumakova<sup>b</sup>, and T. I. Pehk<sup>a</sup>

<sup>a</sup> National Institute of Chemical Physics and Biophysics, Tallinn, Estonia
<sup>b</sup> Shubnikov Institute of Crystallography, Russian Academy of Sciences, Leninskiĭ pr. 59, Moscow, 119333 Russia
e-mail: kiirend@kbfi.ee; kira@ns.crys.ras.ru

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**Abstract**—Nematic lyomesophases with discotic  $(N_{\rm D})$  and cylindrical  $(N_{\rm C})$  micelles in complex multicomponent lyotropic systems based on alkyltrimethylammonium bromide detergents have been identified by the  $^{1}$ H-, and  $^{13}$ C-NMR methods and polarization optical microscopy. The difference in the structures of the  $N_{\rm D}$  and  $N_{\rm C}$  nematic phases is especially pronounced in the  $^{13}$ C-NMR spectra. Addition of chiral dopants to the lyomixture facilitates formation of the  $Ch_{\rm D}$  and  $Ch_{\rm C}$  cholesteric phases. According to the  $^{13}$ C-NMR spectra, the micellar mobility in the cholesteric lyomesophases decreases in comparison with the nematic ones. The alignment of lyocholesterics under the action of an external magnetic field is found.

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#### INTRODUCTION

In 1967, Lawson and Flautt managed to align for the first time the lyotropic phase in a dc magnetic field [1]. The oriented lyotropic phase was formed in a multicomponent mixture composed of sodium decylsulfate, *n*-decyl alcohol, Na<sub>2</sub>SO<sub>4</sub>, and D<sub>2</sub>O. The lyotropic liquid crystal (LC) phases capable of originating in a magnetic field were referred to as lyonematics. Most of such mixtures were synthesized on the basis of alkyl sulfates with the hydrocarbon chain length  $C_{12}$  (12 carbon atoms). Lyonematics with disklike ( $N_D$  and  $N_L$ ) and cylindrical ( $N_C$  and  $N_H$ ) micelles [2-6] and biaxial lyonematics  $(N_b)$  [7] are known. Most often, the orientation of amphiphilic lyonematics was studied experimentally by the <sup>2</sup>H nuclear magnetic resonance (NMR) method [2, 8, 9]. The signal from deuterium nuclei in anisotropic phases is split into a doublet. The value of splitting is proportional to the degree of order S = $\langle P_2(\cos\theta) \rangle$ , where  $\theta$  is the angle between the director of lyotropic LC and the axis of an individual conglomerate (micelle), and the orientational parameter  $P_2(\cos \varphi)$ . The angle  $\varphi$  characterizes the dependence of the orientation of the LC director on the direction of external magnetic field  $H_0$  [8]. Study of lyotropic nematic phases in various amphiphilic systems is promising for the development of a perfect model of biomembranes and bilayers and the investigation of interphase transitions in biosystems. It is known, for example, that a partial macroscopic orientation in magnetic and electric fields is characteristic of the membrane of the ε coli bacteria [10]. Thus, analysis of the formation of lyone-matic phases with various types of micelles in multi-component amphiphilic systems and their alignment under the action of external magnetic, electric, and other fields is now an urgent problem. The application of the <sup>13</sup>C-NMR method to the investigation of lyosystems based on a nonionogenic detergent was reported in [11].

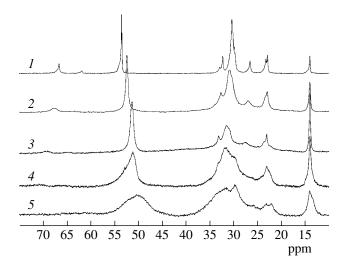
#### **EXPERIMENTAL**

We synthesized multicomponent lyosystems: cetyltrimethylammonium bromide (CTAB I)-n-decyl alcohol-NaBr-D<sub>2</sub>O. The detergent used (Serva) was a mixture of dodecyl- (25 mol % C<sub>12</sub>Me<sub>3</sub>ABr), tetradecyl-(65 mol %  $C_{14}Me_3ABr$ ), and hexadecyl- (10 mol % C<sub>16</sub>Me<sub>3</sub>ABr) trimethylammonium bromide. The composition of CTAB I was analyzed by the <sup>13</sup>C-NMR method on an AMX-500 Bruker spectrometer. In the multicomponent lyotropic mixtures synthesized on the basis of this multidetergent [5], we found the  $N_D$  and  $N_C$ nematic phases. For comparison, a lyomixture based on the detergent CTAB II (Serva), consisting of only hexadecyltrimethylammonium bromide ( $C_{16}Me_3ABr$ ), was prepared. In the ternary lyotropic mixtures, the  $N_{\rm D}$  and  $N_{\rm C}$  nematic phases similar to those observed in [12] (where the lyomixtures were prepared on the basis of cetyldimethylethylammonium bromide) were formed. In our case, the nematic phases formed in the lyomixtures with the detergent CTAB II were inhomogeneous: Composition of synthesized mixtures; identified mesophases; types of textures determined by polarization-optical microscopy; and the data on the characteristic quadrupole splitting of the <sup>2</sup>H-NMR signal of water. Samples 5 and 6 contained cetyldimethylethylammonium bromide detergent [12]

Sample no.	Phase	Texture	Temper- ature, K	$\Delta v_D$ , Hz	Refer- ence	Mixture composition, wt %					
						CTAB I	CTAB II	1-dec- anol	NaBr	D <sub>2</sub> O	Chiral dopant
1	Mi		297			32.36		3.88	3.06	60.70	
2	$N_{\rm C}$	marble	286	14.4		32.36		3.88	3.06	60.70	
3	$N_{ m D}$	schlieren	297	20.0		27.99		6.84	4.08	61.09	
4	$N_{ m D}$	schlieren	297	20.0		27.95		4.08	6.80	61.17	
5	Mi		297		[12]		22.95	2.05		75.00	
6	$Ch_{\mathbb{C}}$	homogeneous planar	297		[12]		22.95	2.05		75.00	1.56 wt % cholesterol
7	$Ch_{\mathrm{D}}$	schlieren	297			26.45		3.53	3.71	61.89	4.42 wt % L-arginine
8	$L_{\alpha}$	confocal	297	117.0			32.13	8.75		59.12	
9	homogeneous phase was not formed						32.36	3.88	3.06	60.70	

the presence of the two phases was always observed in anisotropic systems at a fixed temperature corresponding to the presence of nematic lyomesophases.

The detergents CTAB I and CTAB II (Serva), n-decyl alcohol (reagent grade), and NaBr (reagent grade), cholesterol (Reakhim, Russia), and *L*-arginine (Reanal, Hungary) were used without additional purification. Deuterated water contained more than 99.8% D<sub>2</sub>O. The presence of the detergent CTAB I in the lyosystem synthesized by mixing individual components in a certain weight ratio (see table) and further centrifugation of the mixture in sealed glass ampoules (8–



**Fig. 1.**  $^{13}$ C-NMR spectra of samples 1, 2, and 3: (1) micellar phase, (2)  $N_{\rm C}$  phase, (3)  $N_{\rm D}$  phase, (4)  $L_{\alpha 1}$  lamellar phase, and (5)  $L_{\alpha}$  phase formed in the 32.12 wt % CTAB II–8.75 wt % n-decanol–59.12 wt % D<sub>2</sub>O lyomixture in the absence of NaBr salt.

9 mm in diameter and narrowed to 1-2 mm in the center) facilitated easy formation of nematic lyomesophases. Lyonematics in the mixtures with the detergent CTAB II were much harder to form. The mixtures were always heated to the isotropic-phase temperature 60-70°C, stirred for 60-100 min, cooled to room temperature, and centrifugated to obtain a homogeneous sample. Details of the investigation technique were reported in [4–6, 13]. The <sup>2</sup>H- and <sup>13</sup>C-NMR spectra were recorded on a pulsed Bruker spectrometer at a frequency of 76.8 MHz (with a polarizing magnetic field of 11.7 T) at  $30 \pm 0.5$  °C. The magnetic field was applied parallel to the long axis of the ampoule. The nematic  $N_{\rm D}$ and  $N_C$  phases were identified by analyzing the <sup>2</sup>H- and <sup>13</sup>C-NMR spectra and the polarization optical microscopy data. Nematic mesophases with discotic micelles  $N_{\rm D}$  exhibit a schlieren texture in thin layers (thinner than 100 µm) and capillaries. In thick capillaries (>100–120 µm), this texture becomes homeotropic. The nematic  $N_{\rm C}$  phases in capillaries exhibit either marble or planar texture.

### **RESULTS**

The composition of the amphiphilic mixtures prepared on the basis of the detergents CTAB I and CTAB II, the phases observed in these lyomixtures and identified by the <sup>2</sup>H- and <sup>13</sup>C-NMR methods and polarization optical microscopy, and the textures characteristic of these phases are listed in table.

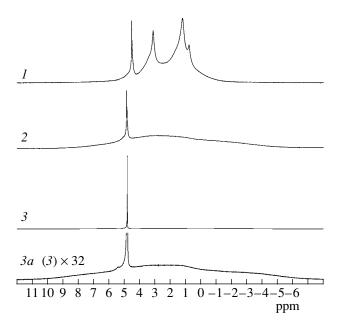
It was shown previously [4, 5, 13] that the splitting  $\Delta v_D$  in the <sup>2</sup>H-NMR spectra of anisotropic lyonematic  $N_C$  phases with cylindrical micelles is less than the splitting for the  $N_D$  phases with discotic micelles. This distinction can also be seen from the data in table. The

difference in the structure of the  $N_{\rm D}$  and  $N_{\rm C}$  nematic lyomesophases is pronounced in the <sup>2</sup>H- and <sup>13</sup>C-NMR spectra. As an example, Fig. 1 shows the <sup>13</sup>C-NMR spectra of the CTAB I multicomponent mixture, n-decyl alcohol-NaBr-D<sub>2</sub>O (table, sample 1). The highest mobility of the carbon atoms of the N-methyl groups of the detergent molecule is inherent in the nematic  $N_C$  phase (Fig. 1, curve 2). The <sup>13</sup>C-NMR spectra of the  $N_{\rm C}$  phase are similar to those of the micellar Miphases at an identical composition of the mixture components (Fig. 1, spectra 1, 2; table, sample 1). It can be seen that the mobility of the carbon atoms of the terminal group CN(CH<sub>3</sub>)<sub>3</sub> of the detergent molecule is most hindered in the  $N_D$  phase, although the  $N_D$  phases are less viscous than the  $N_{\rm C}$  phases. In addition, Fig. 1 shows the  ${}^{13}\text{C-NMR}$  spectra of the  $L_{\alpha 1}$  lamellar phase (curve 4) observed in the synthesized lyomixture with the composition 32 wt % CTAB I-9.25 wt % NaBr-5.54 wt % *n*-decanol–52.31 wt % D<sub>2</sub>O (the value of splitting  $\Delta v_D$  = 68 Hz at 297 K) [13] and the  $L_{\alpha}$  phase formed in the 32.12 wt % CTAB II-8.75 wt % *n*-decanol-59.12 wt %  $D_2O$  mixture in the absence of NaBr ( $\Delta v_D = 117$  Hz).

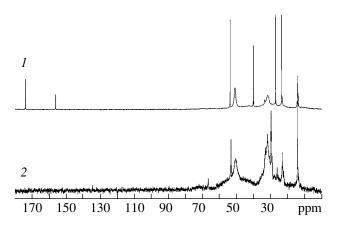
A comparison of the <sup>13</sup>C-NMR data with the composition of the samples under study showed that the highest mobility of the carbon atoms of the long molecular chain of the detergent in LC phases is observed in the presence of the NaBr salt in the lyomixture. In the absence of NaBr, all mixtures form lyophases, which, according to the spectra, are characterized by a lower mobility of the carbon atoms in the methyl chain of the detergent molecules. The <sup>13</sup>C-NMR spectra of lamellar phases (Fig. 1, curves 4, 5) illustrate the influence of the NaBr salt on the formation of polymorphic lamellar phases [13]. This influence manifests itself in the different mobilities of the carbon atoms in the alkyl chain of the detergent molecules. Analysis of the <sup>1</sup>H- (Fig. 2) and <sup>13</sup>C-NMR spectra and the polarization-optical microscopy data allowed us to identify the  $N_D$  and  $N_C$ nematic lyomesophases formed in the synthesized multicomponent mixtures based on the detergent CTAB I (see table).

The  $^{1}$ H-NMR spectra of the  $N_{\rm C}$  phase (Fig. 2, curve 2; table, sample 2) and the  $N_{\rm D}$  phase (sample 3, spectra 3, 3a; the latter is enlarged by a factor of  $\times$ 32) show only the peaks of free-water protons. The  $^{1}$ H-NMR spectrum of the micellar phase (spectrum I; table, sample 1) contains peaks from both free-water protons and protons of the alkyl chain of the detergent.

Addition of 4.42 wt % L-arginine to sample 4 ( $N_{\rm D}$  phase) led to the formation of a cholesteric (Ch) lyomesophase with discotic micelles ( $Ch_{\rm D}$ ) in the lyomixture. The  $^{13}$ C-NMR spectrum of the  $Ch_{\rm D}$  phase (Fig. 3, spectrum I; sample 7) indicates that the motion of the carbon atoms of the alkyl chain of the detergent molecules is more hindered in this phase in comparison with the cholesteric phase with cylindrical micelles  $Ch_{\rm C}$  (Fig. 3,



**Fig. 2.**  $^{1}$ H-NMR spectra: (1) micellar phase of sample 1, (2)  $N_{\rm C}$  phase of sample 2, and (3, 3a)  $N_{\rm D}$  phase of sample 3 (spectrum 3a is spectrum 3 enlarged by a factor of 32).



**Fig. 3.** <sup>13</sup>C-NMR spectra of cholesteric phases: (1)  $Ch_D$  phase, T = 297 K, sample 7; (2)  $Ch_C$  phase, T = 297 K, sample 6.

spectrum 2), which was observed as a result of addition of 1.56 wt % cholesterol to the lyomixture of sample 5 (table). The  $N_{\rm C}$  and  $Ch_{\rm C}$  phases are more viscous than the  $N_{\rm D}$  and  $Ch_{\rm D}$  phases. However, as can be seen from the NMR spectra, the motion of the carbon atoms of the alkyl chain of the detergent molecules is most hindered in the  $N_{\rm D}$  and  $Ch_{\rm D}$  phases. An analysis of the  $^{13}{\rm C}$  NMR spectra of the  $N_{\rm D}$  and  $N_{\rm C}$  nematic phases and the  $Ch_{\rm D}$  and  $Ch_{\rm C}$  cholesteric phases showed that the micellar mobility in the cholesteric lyophases is lower than in the nematic ones. Polarization optical measurements revealed a typical disordered cholesteric texture of the  $Ch_{\rm D}$  phase (fingerprint texture; Fig. 4a, the bottom part of the capillary), which can be aligned by a magnetic





**Fig. 4.** Textures of cholesteric lyomesophases: (a)  $Ch_{\rm D}$  phase of sample 7 and (b)  $Ch_{\rm C}$  phase of sample 6. The bottom part shows the unaligned texture regions. In the top part, the texture regions are aligned by the external magnetic field H=11.7 T applied parallel to the long axis of the capillary. The capillary thickness is  $d=100~\mu{\rm m}$ ; the magnification is (a)  $\times 250$  and (b)  $\times 100$ .

field (H = 11.7 T) applied parallel to the sample's long axis (Fig. 4a, the top part of the capillary). The  $Ch_C$  cholesteric phase forms a defect planar structure (Fig. 4b, the bottom part of the capillary) which likewise is aligned under the action of a magnetic field (Fig. 4b, the top part of the capillary). The  $\lambda^+$  and  $\tau^+$  disclinations, which are typical of cholesterics, can be seen in the oriented texture of the  $Ch_C$  phase.

#### **CONCLUSIONS**

Thus, the analysis of the  $^{1}\text{H-}$ ,  $^{2}\text{H-}$ , and  $^{13}\text{C-NMR}$  spectra and the polarization-optical microscopy data obtained for the mesophases formed in the multicomponent mixtures synthesized on the basis of the CTAB detergent allowed us to identify nematic lyophases with discotic ( $N_{\rm D}$ ) and cylindrical ( $N_{\rm C}$ ) micelles. Addition of

a small amount of chiral dopants (L-arginine and cholesterol) to these lyophases leads to the formation of the  $Ch_D$  and  $Ch_C$  phases. It was found from the  $^{13}$ C-NMR spectra that the micellar mobility in cholesteric phases decreases in comparison with the nematic lyomesophases. The  $^{13}$ C NMR method allows one to obtain information about the dynamics of the carbon atom motion in the alkyl chain of the detergent molecules in the lyomesophases with various structures.

#### REFERENCES

- K. D. Lawson and T. J. Flautt, J. Am. Chem. Soc. 89, 5489 (1967).
- 2. B. J. Forrest and W. Reeves, Chem. Rev. 89, 1 (1981).
- L. Q. Amaral, O. Santin Filho, G. Taddei, et al., Langmuir 13 (19), 5016 (1997).
- E. O. Kiĭrend, S. P. Chumakova, and T. I. Pekhk, Kristallografiya 31 (4), 732 (1986) [Sov. Phys. Crystallogr. 31, 432 (1986)].
- E. O. Kiĭrend, S. P. Chumakova, and N. R. Ivanov, Kristallografiya 38 (6), 201 (1993) [Crystallogr. Rep. 38, 817 (1993)].
- L. Q. Amaral, C. A. Pimentel, V. R. Tavares, et al., J. Chem. Phys. 79, 2940 (1979).
- 7. Y. Galerne, Mol. Cryst. Liq. Cryst. 165, 131 (1988).
- L. W. Reeves and A. S. Tracey, J. Am. Chem. Soc. 79, 5729 (1975).
- J. Charvolin and B. Mely, Mol. Cryst. Liq. Cryst. 41, 209 (1978).
- J. Soeling, F. Borle, and T. A. Cross, Biochem. Biophys. Acta 814, 195 (1985).
- L. Q. Zheng, M. Suzuki, and T. Inoue, Langmuir 18, 1991 (2002).
- 12. H. D. Dorfler, Adv. Colloid. Interface Sci. **98**, 285 (2002).
- E. O. Kiĭrend, S. P. Chumakova, and T. I. Pehk, Kristallografiya 47 (5), 914 (2002) [Crystallogr. Rep. 47, 849 (2002)].

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